

Spectroscopy of Mass/Charge Selected Cations and Anions in Helium Droplets

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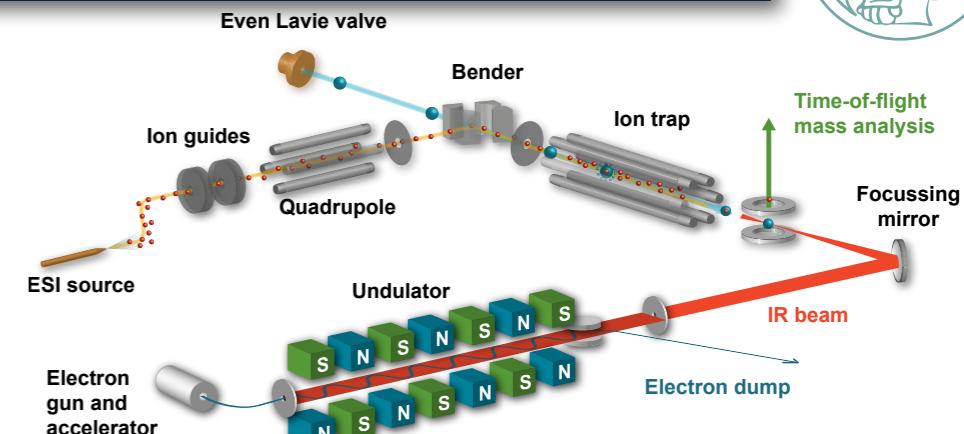


Outline



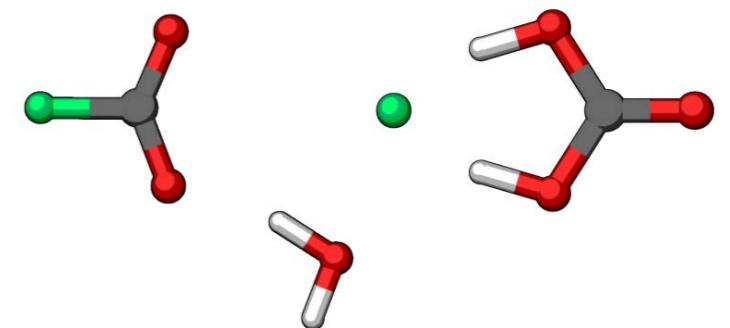
IR Spectroscopy in He Nanodroplets

- m/z selected cations and anions



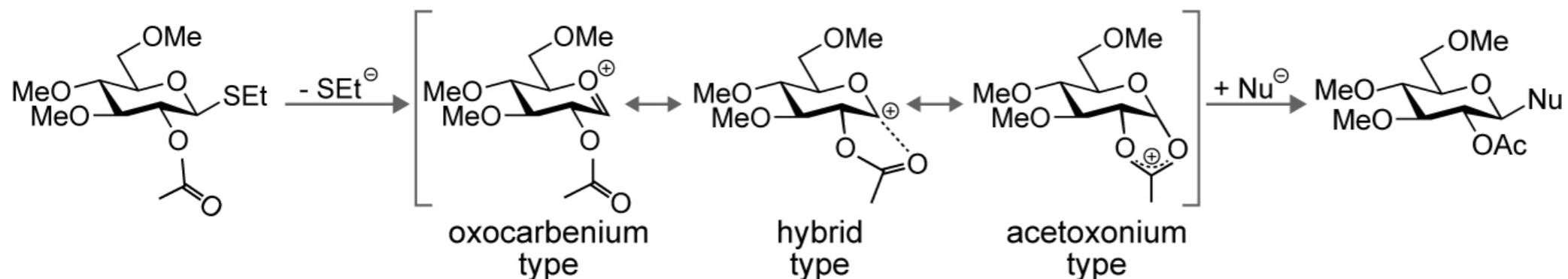
Fluoride Chemistry

- F⁻, and its reactivity towards CO₂ and H₂O



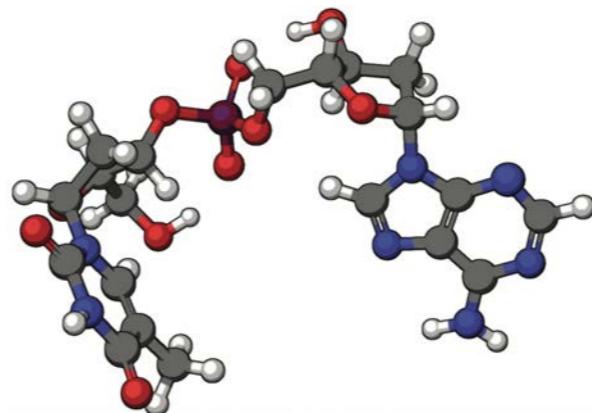
Glycosyl Cations

- Characterisation of an important reaction intermediate



Dinucleotide dTpda

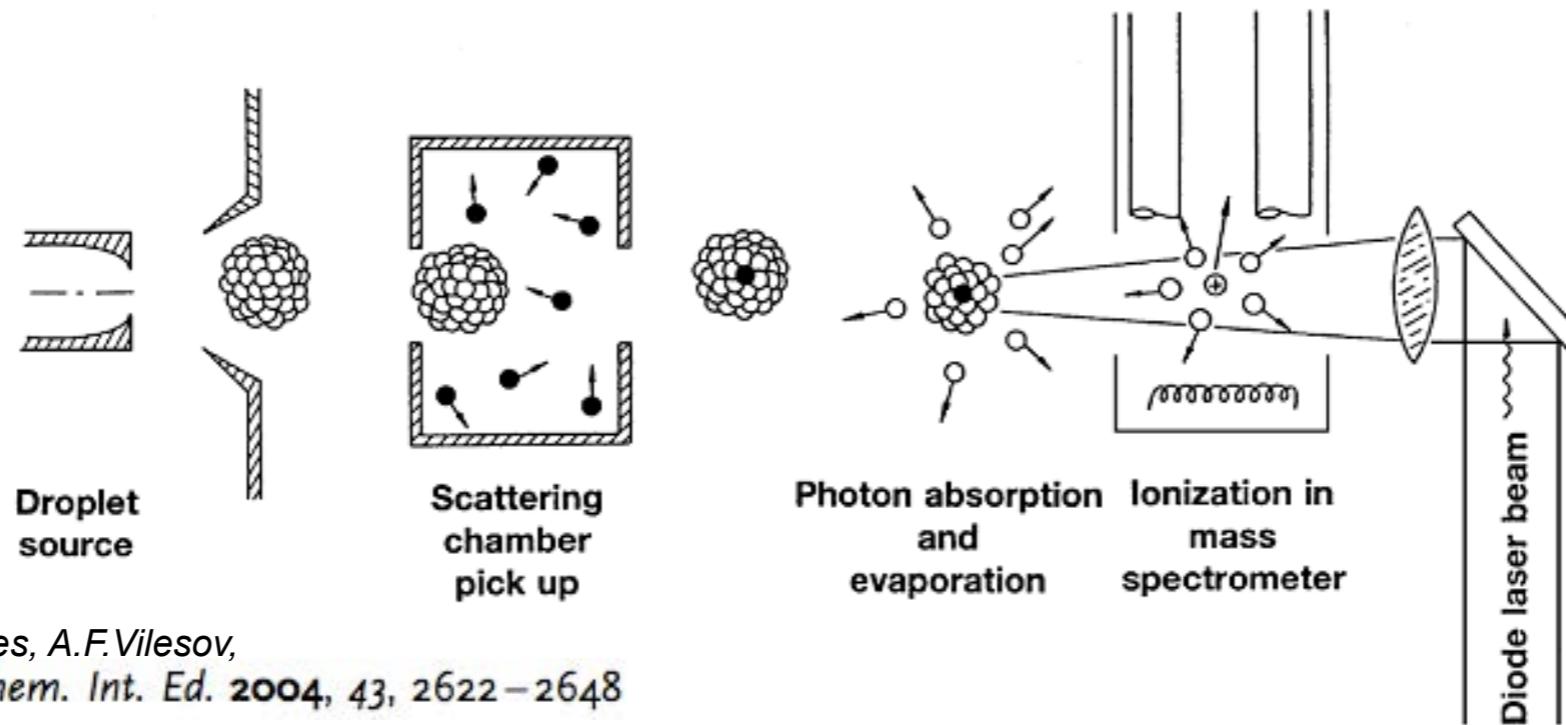
- Probing thermal equilibria



Pickup by liquid helium droplets

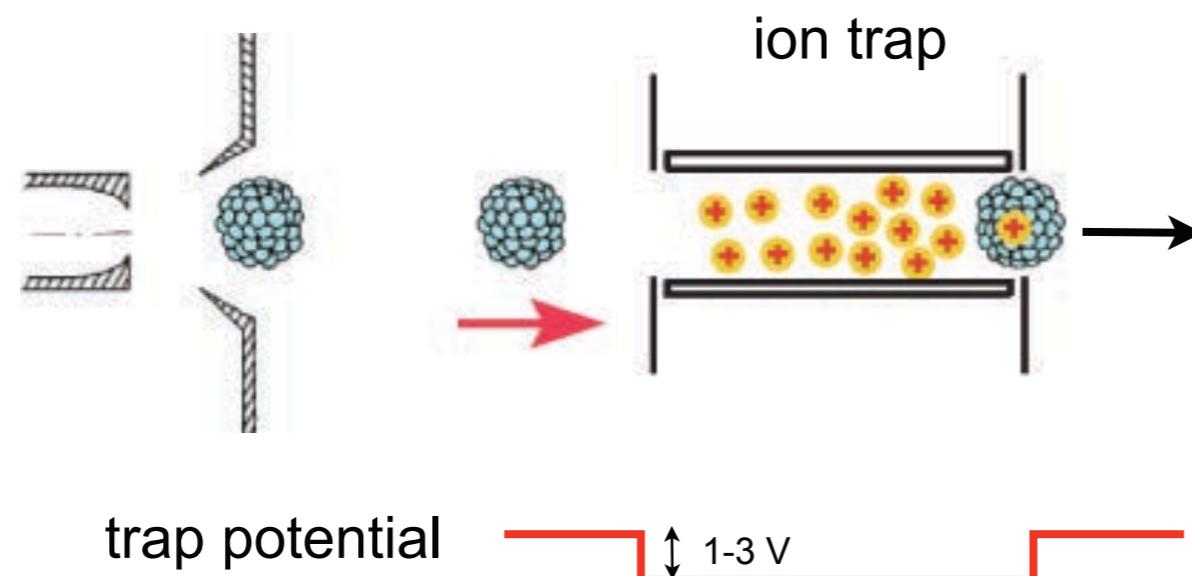


Pick-up of neutral species from a gas cell

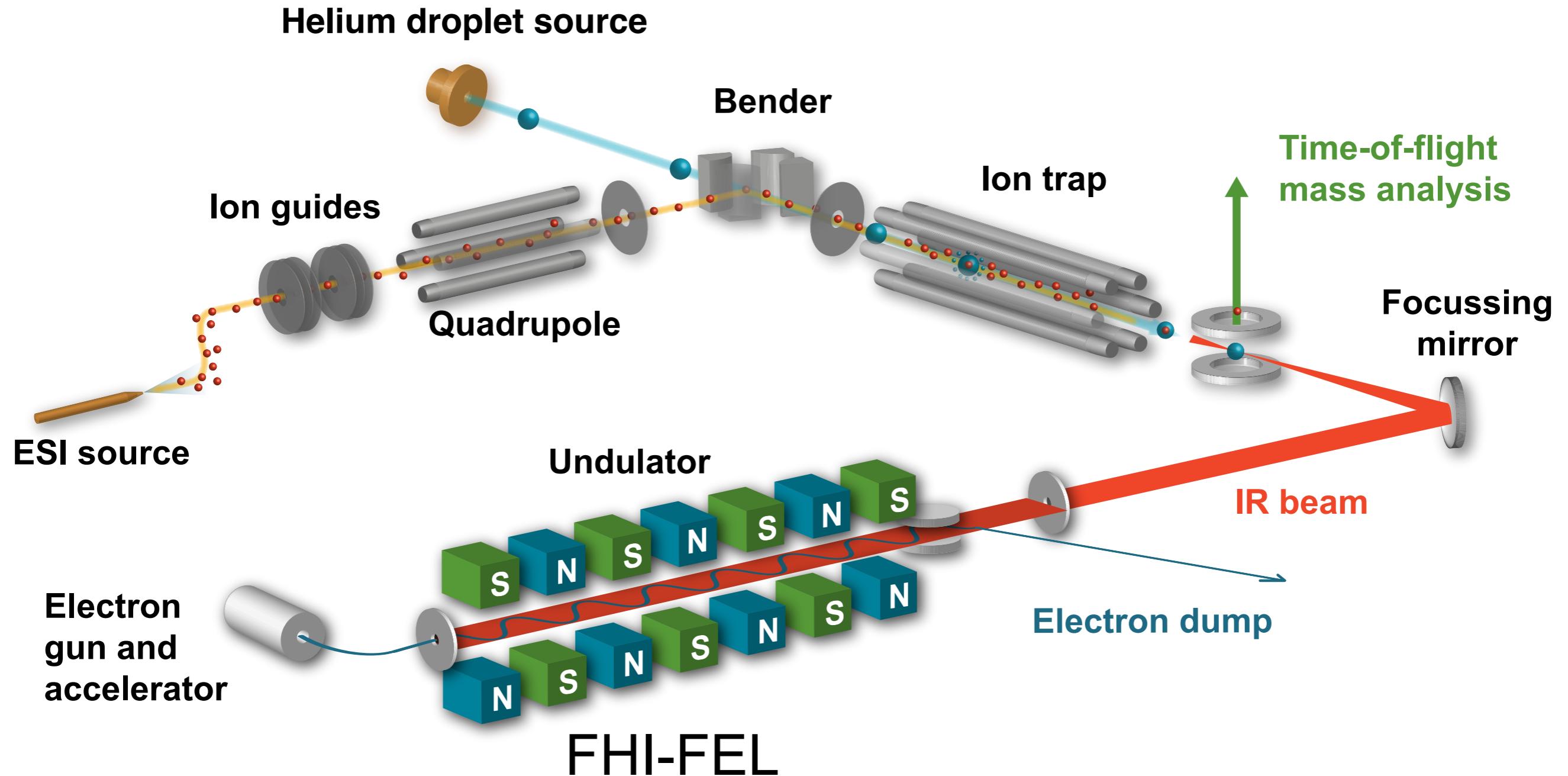


J.P. Toennies, A.F. Vilesov,
Angew. Chem. Int. Ed. 2004, 43, 2622–2648

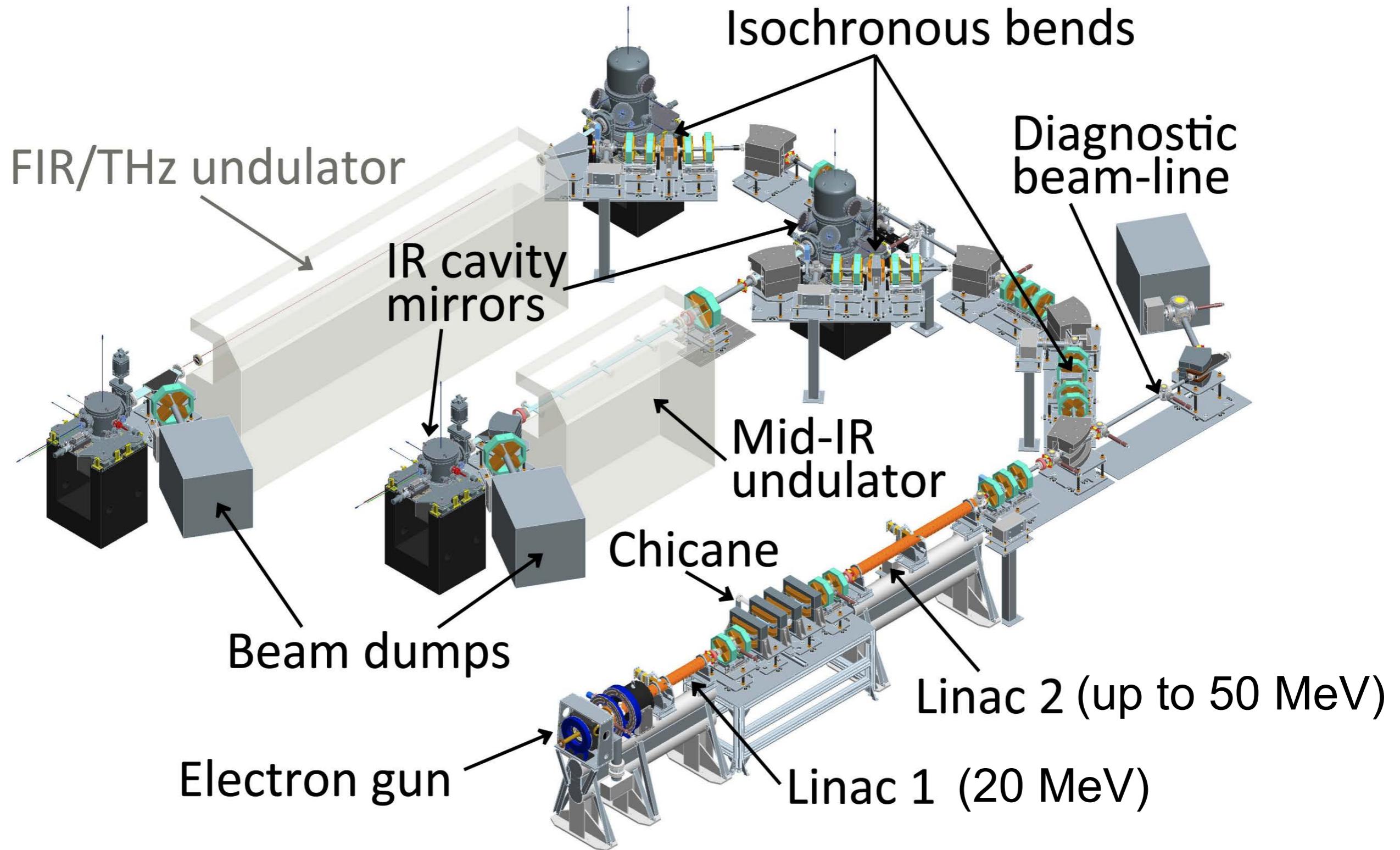
Pick-up of ions from a linear ion trap



Liquid helium droplet setup



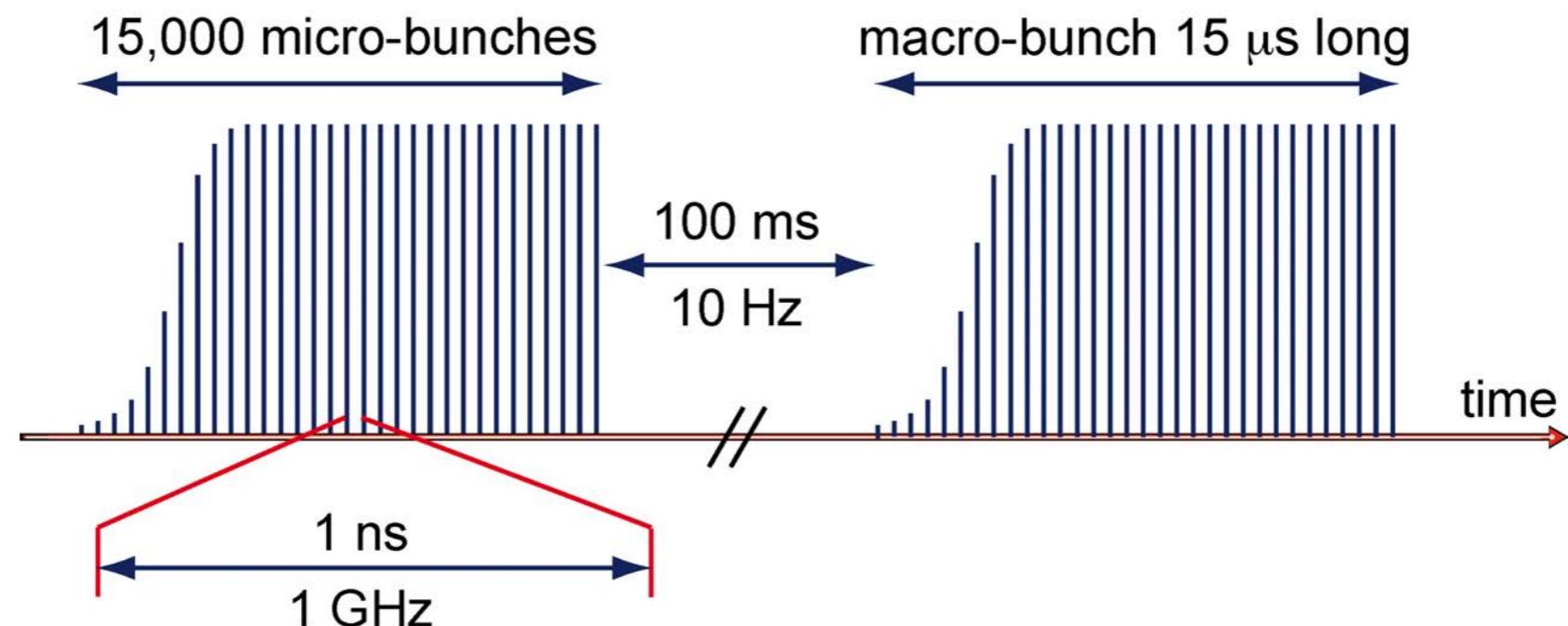
The FHI FEL



Specifications



The time structure of IR output given by electrons:
micro-pulses and macro-pulses



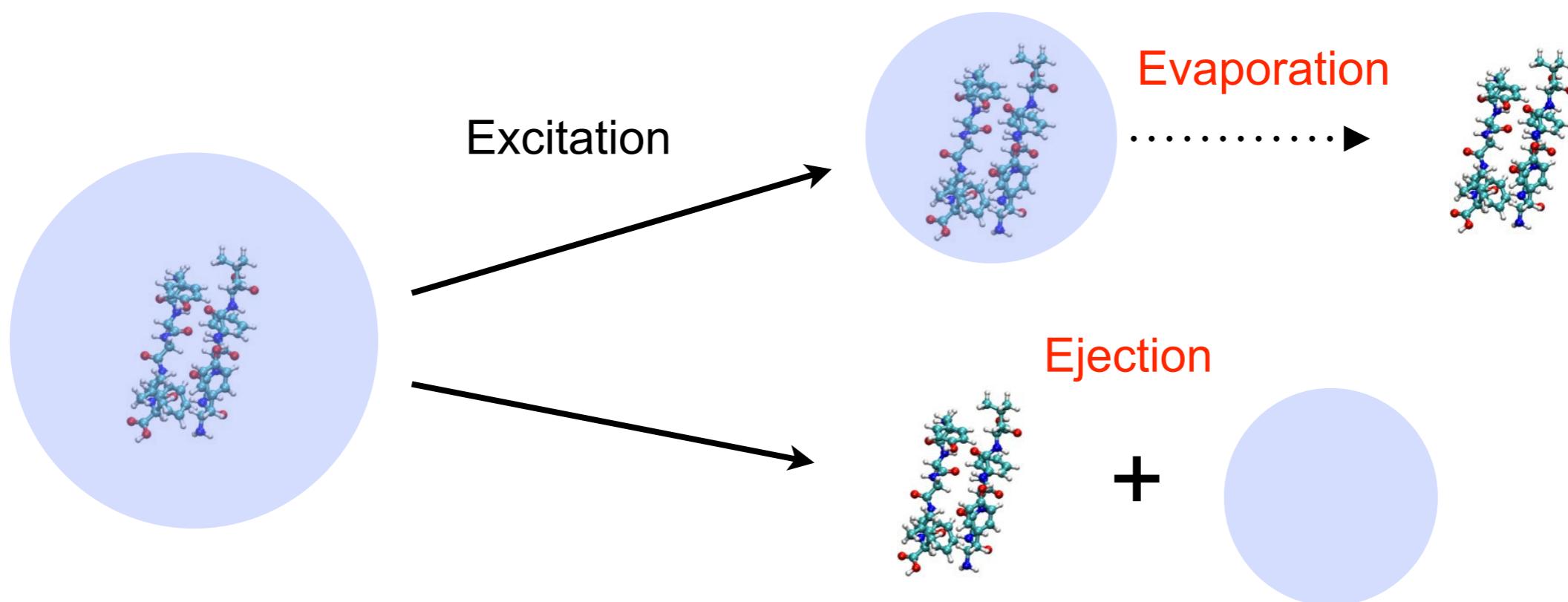
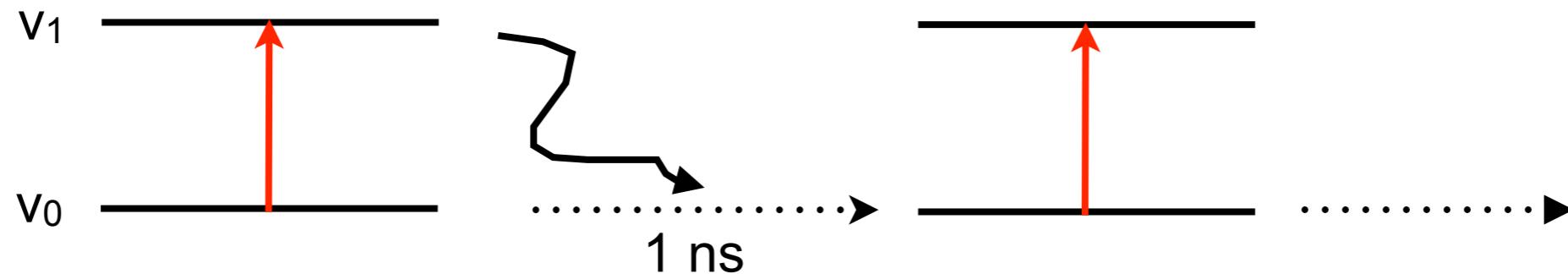
IR output:

- macro-pulse length > 10 μ s
- > 100 mJ / macro-pulse
- 10 – 20 μ J / micro-pulse
- micro-pulse length 0.3 – >5 ps
- FT-limited bandwidth: 0.3 – 5% of central frequency

Action spectroscopy using helium droplets



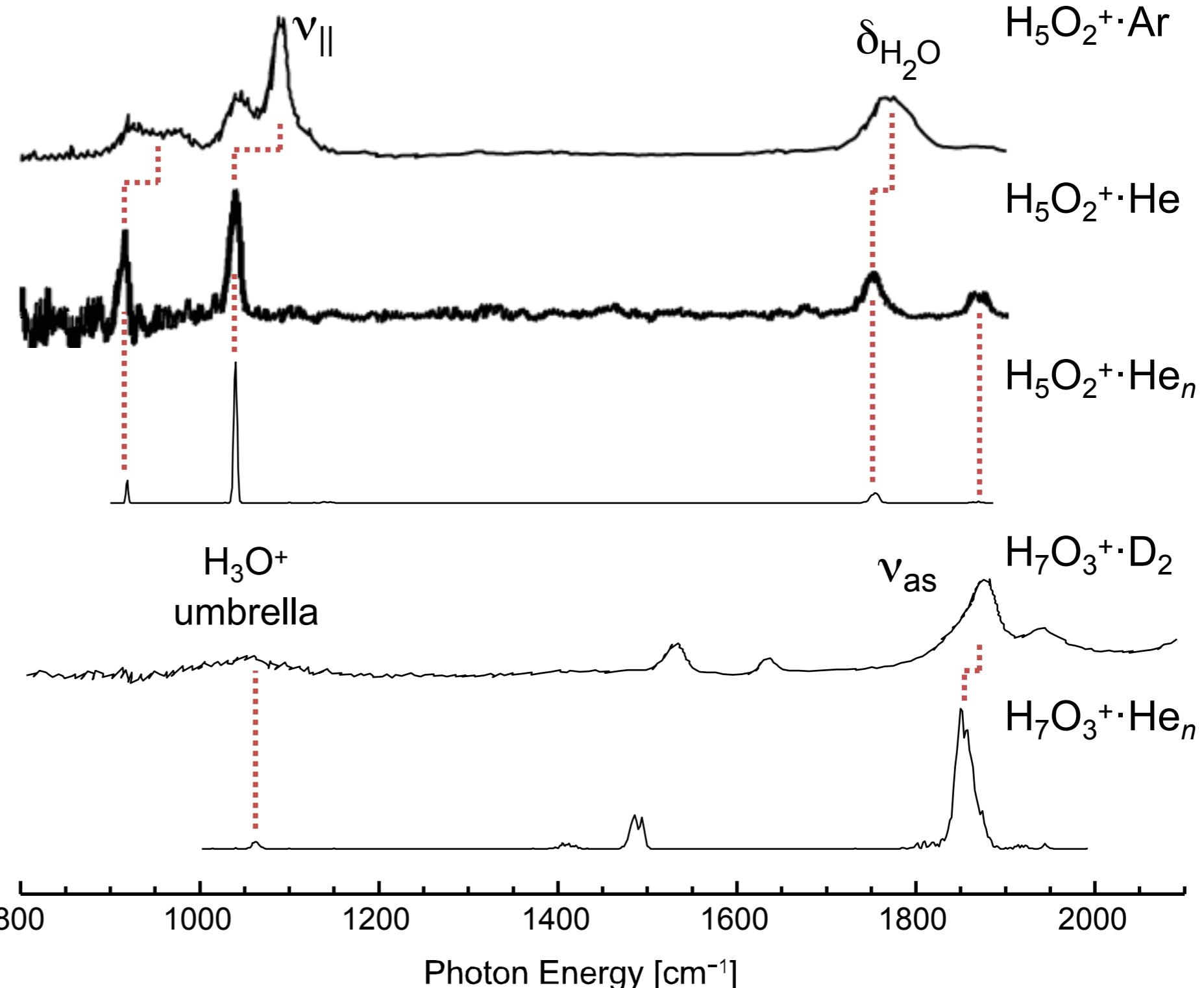
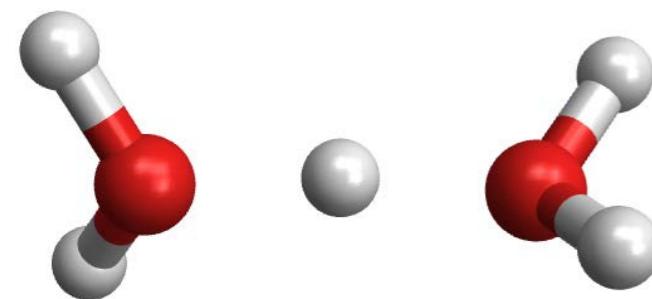
Multiple photon excitation in a macropuls



Protonated Water Clusters



H_5O_2^+ (Zundel):



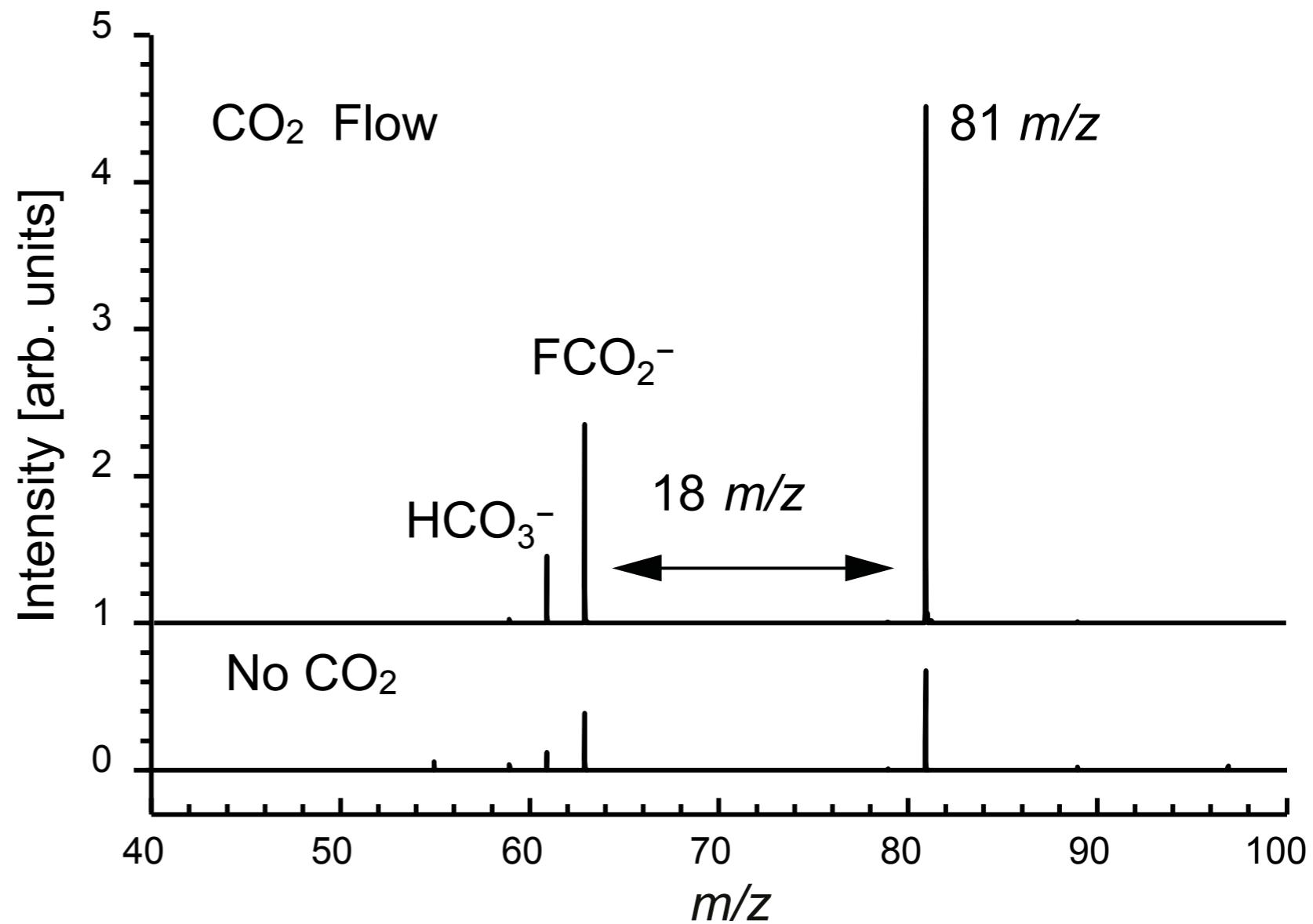
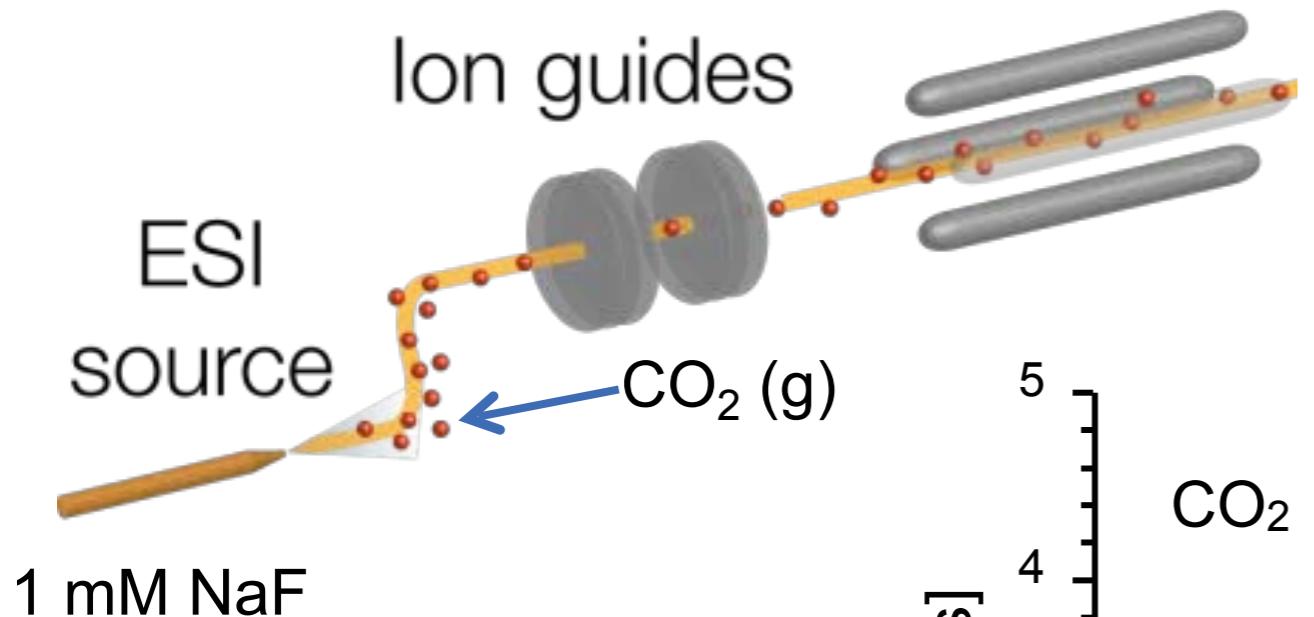
C. J. Johnson *et al.*, *J. Chem. Phys.* **2014**, *140*, 221101.

C. H. Duong *et al.*, *J. Phys. Chem. Lett.* **2017**, *8*, 3782-3789.

Action spectroscopy using helium droplets



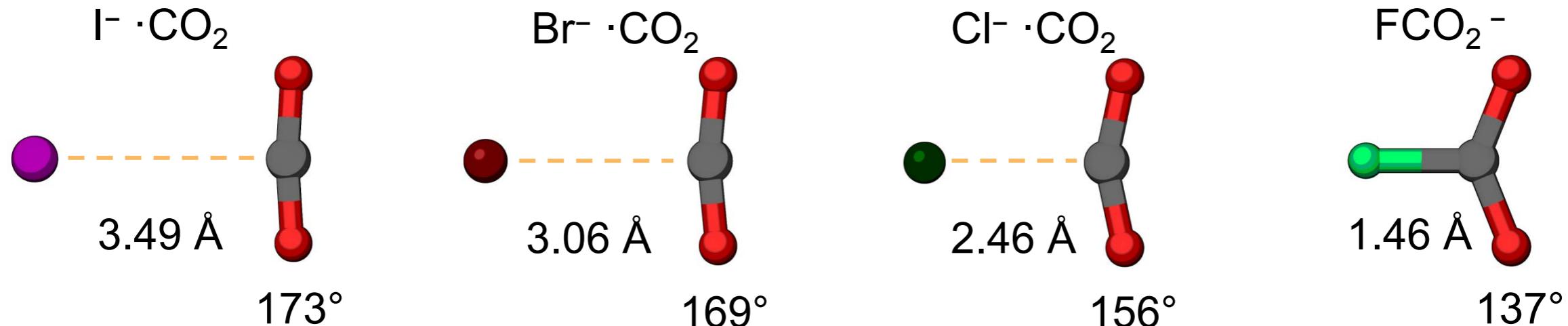
PRODUCTION OF FLUOROFORMATE



Action spectroscopy using helium droplets



HALIDE-CO₂ COMPLEXES

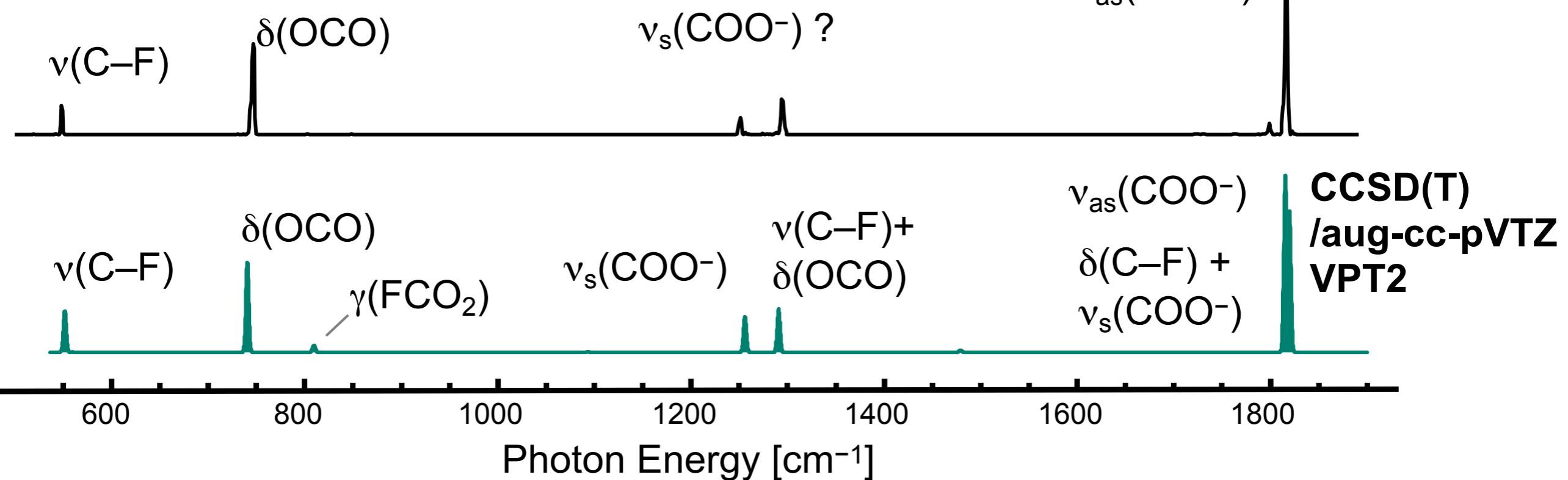
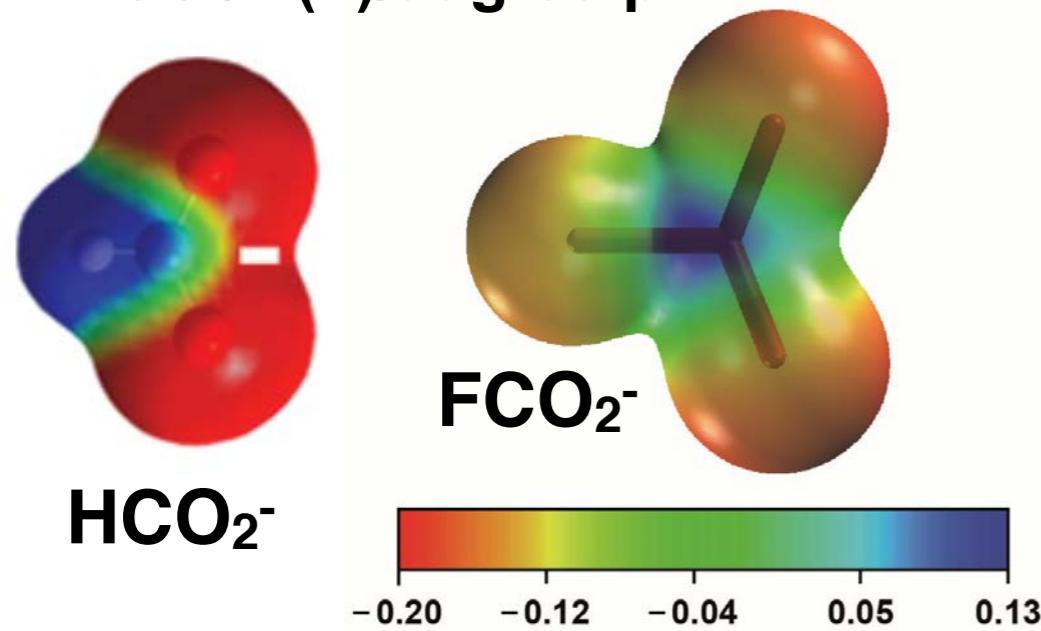


J. M. Weber, H. Schneider, *J. Chem. Phys.* **2004**, *120*, 10056-10061.

IR Spectroscopy of Fluoroformate



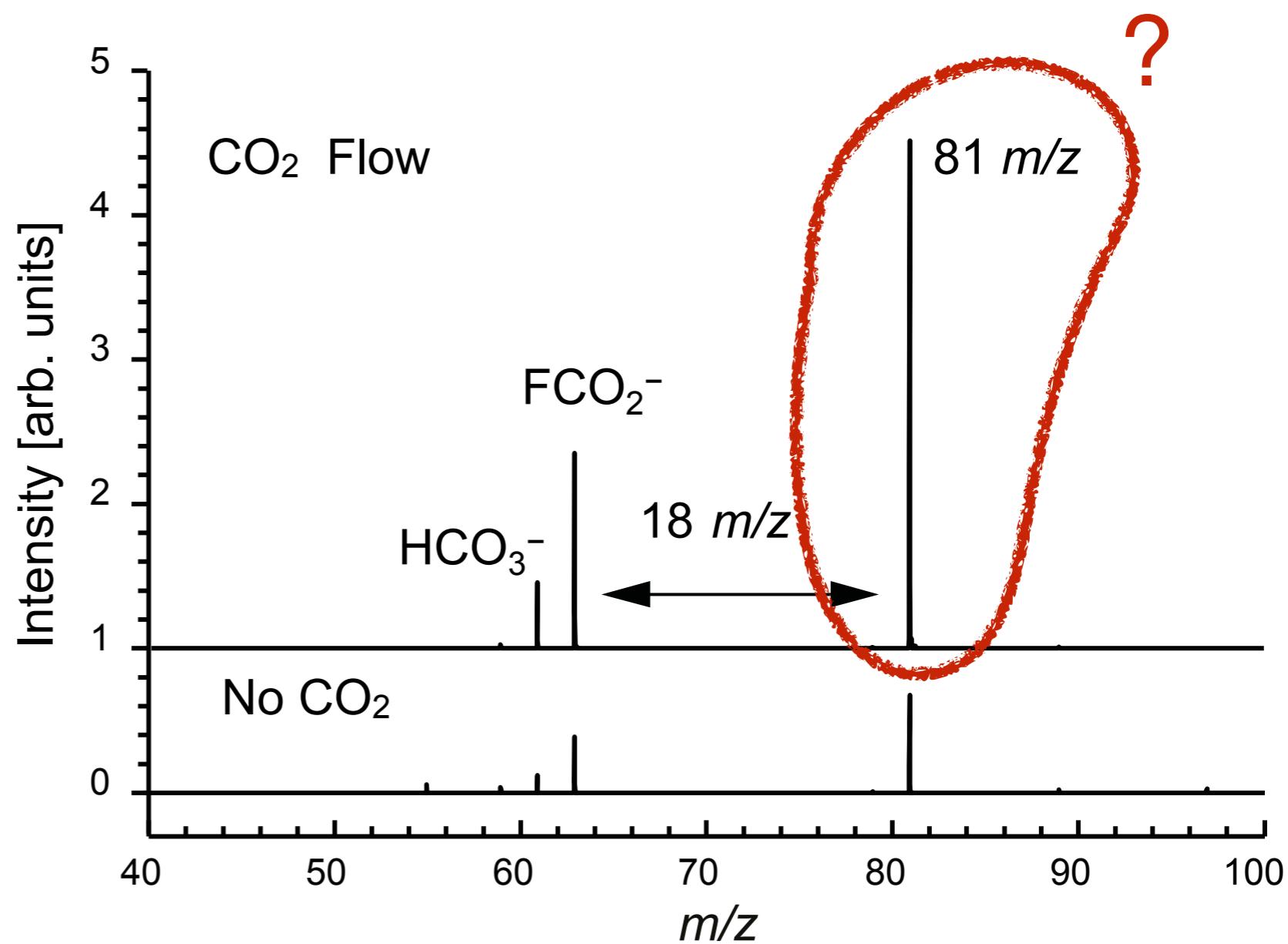
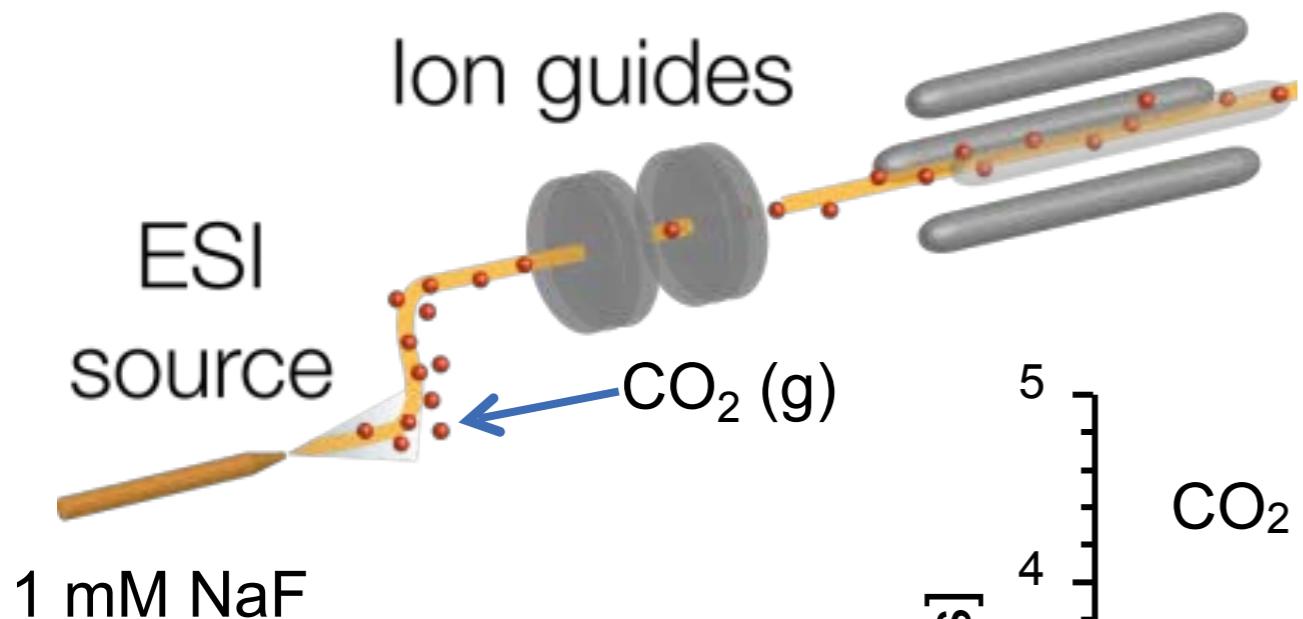
CCSD(T)/aug-cc-pVTZ



IR Spectroscopy of Fluoroformate



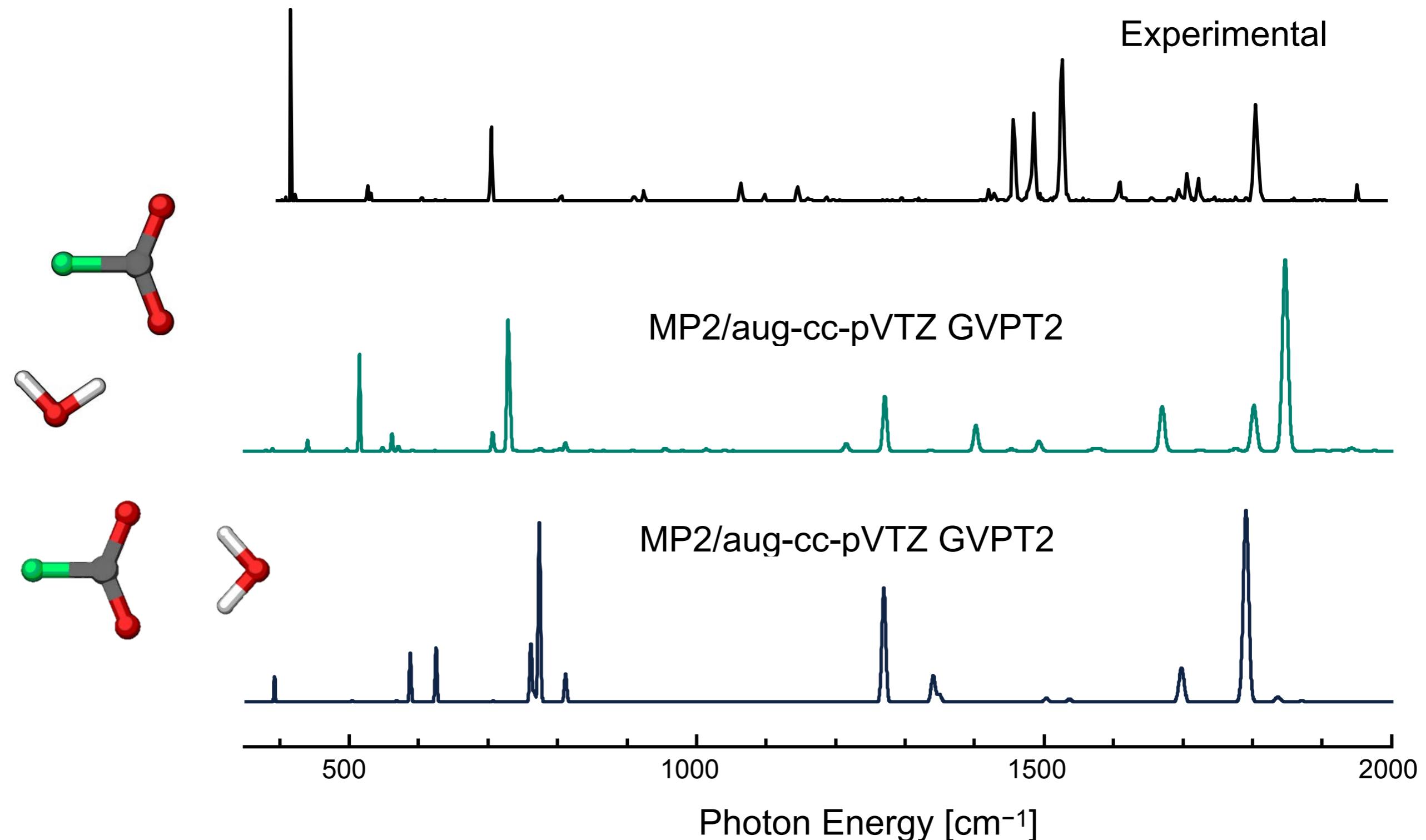
PRODUCTION OF FLUOROFORMATE



IR Spectroscopy of Fluoroformate



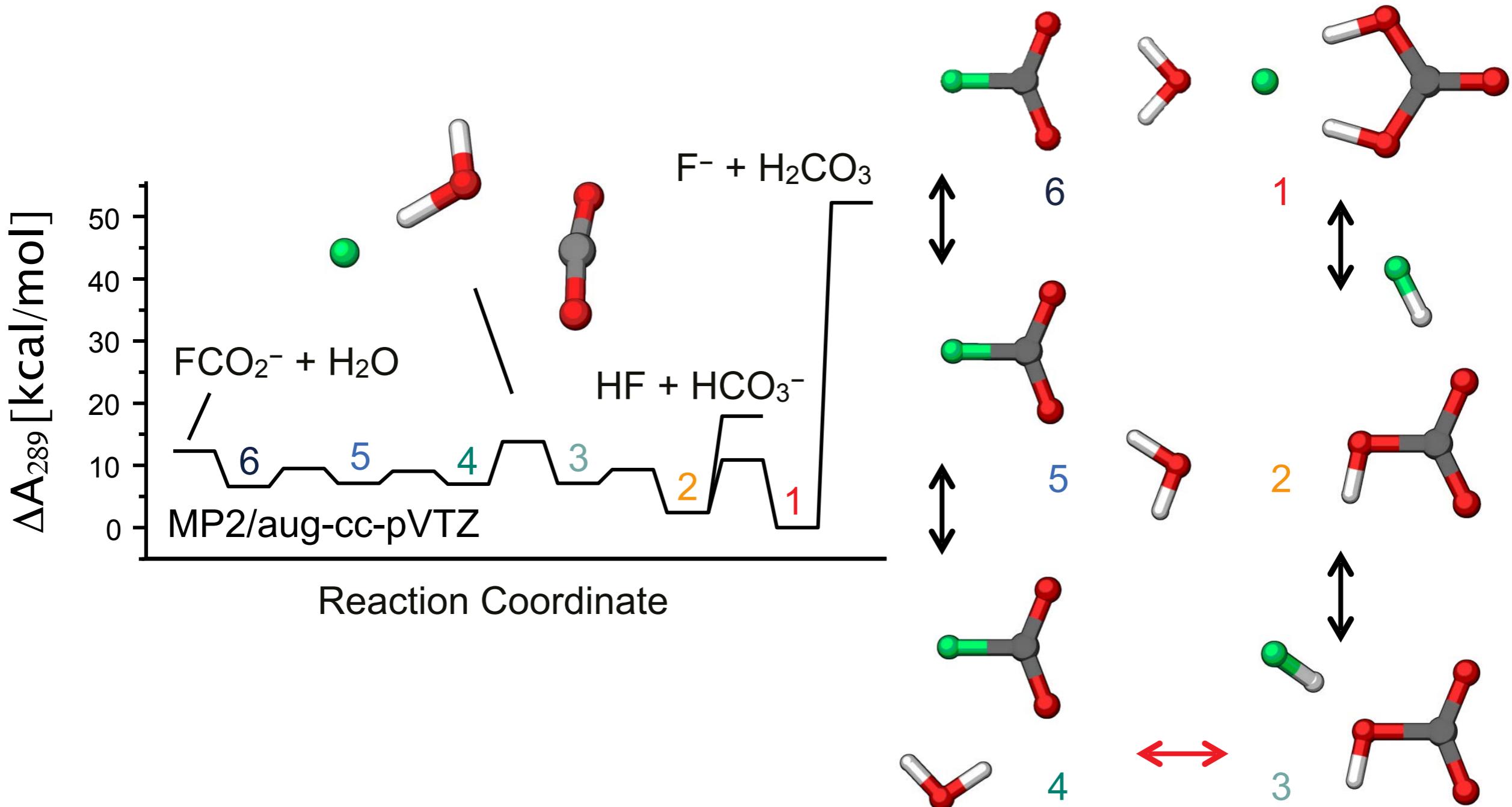
IR SPECTROSCOPY OF $\text{FCO}_2^- + 18$



Action spectroscopy using helium droplets



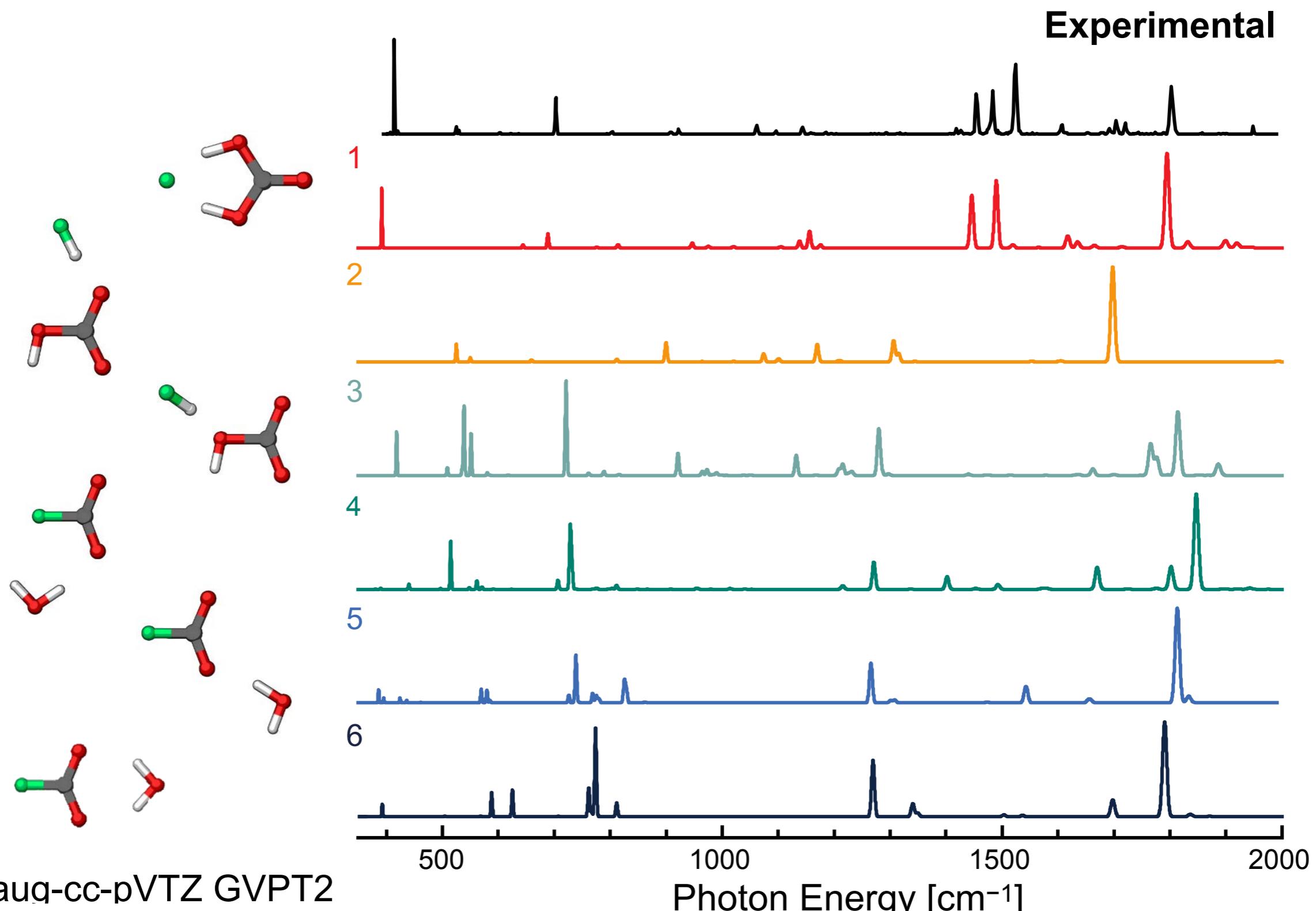
REACTION COORDINATE OF $\text{FCO}_2^- + \text{H}_2\text{O}$



Action spectroscopy using helium droplets



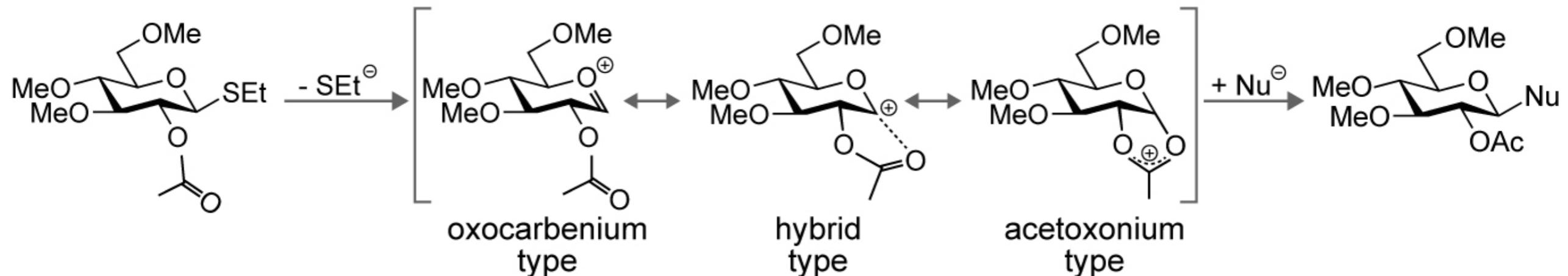
IR SPECTROSCOPY OF $\text{FCO}_2^- + 18$



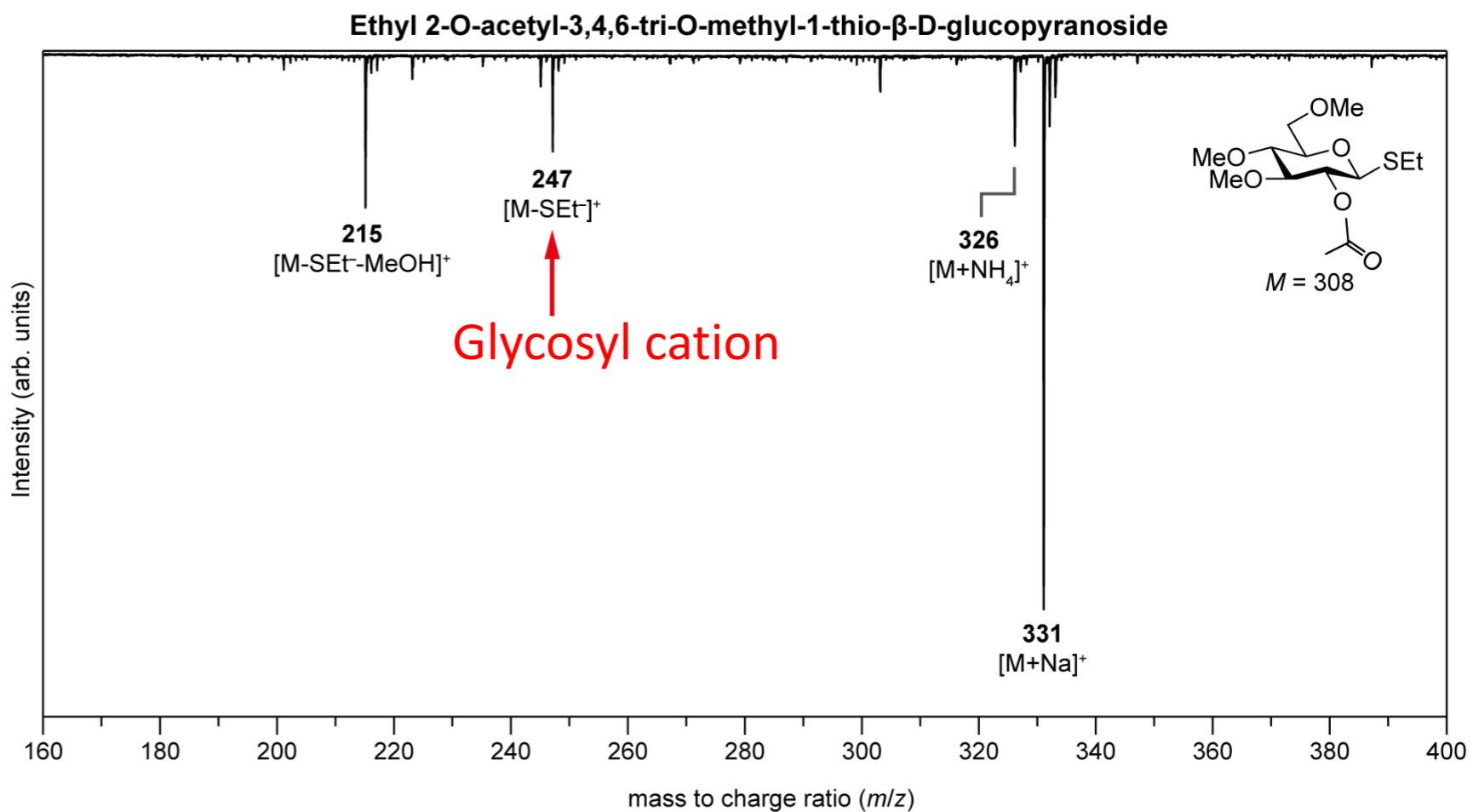
Probing the Structure of Glycosyl Cations



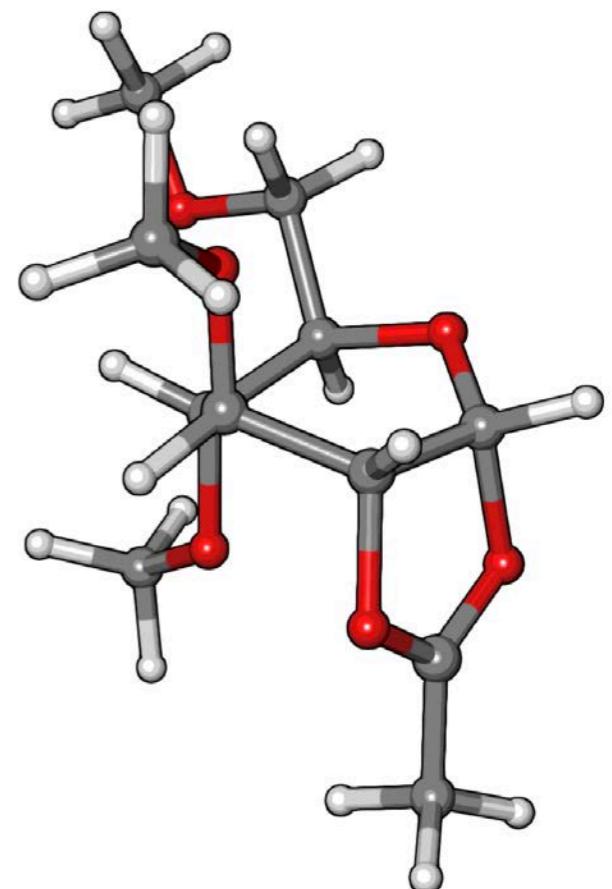
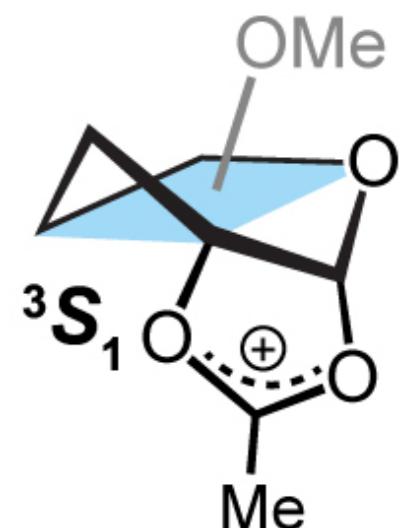
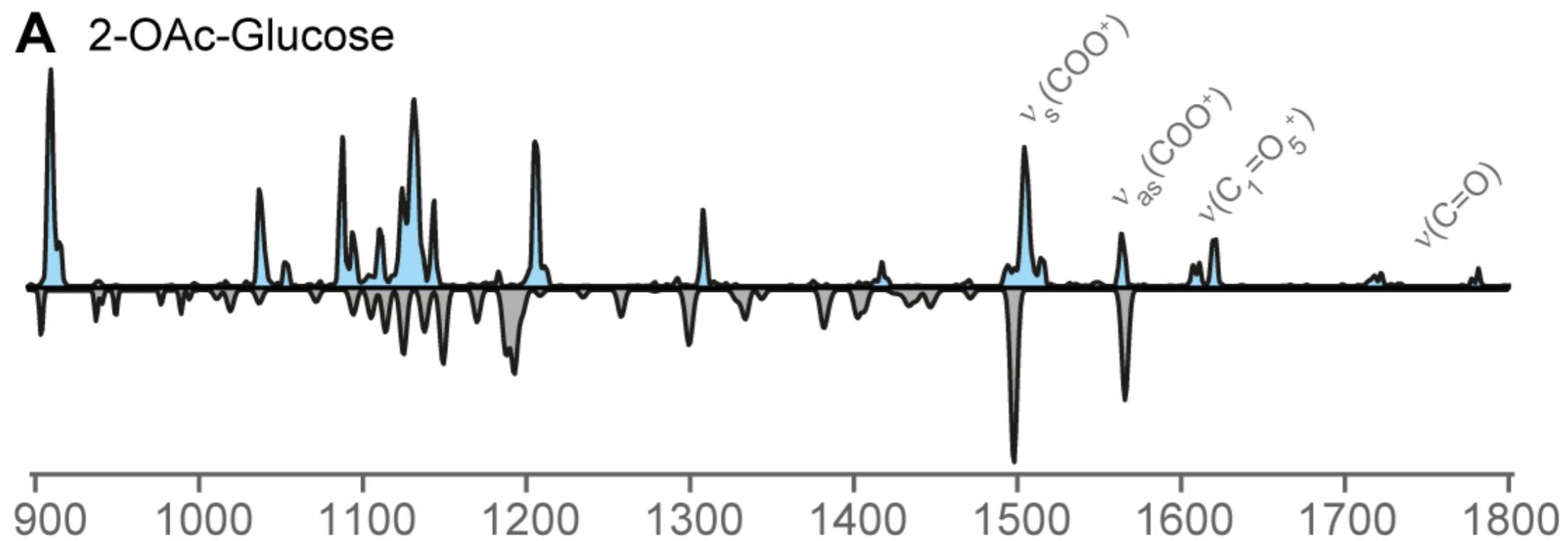
Stereoselective glycosylation reactions:



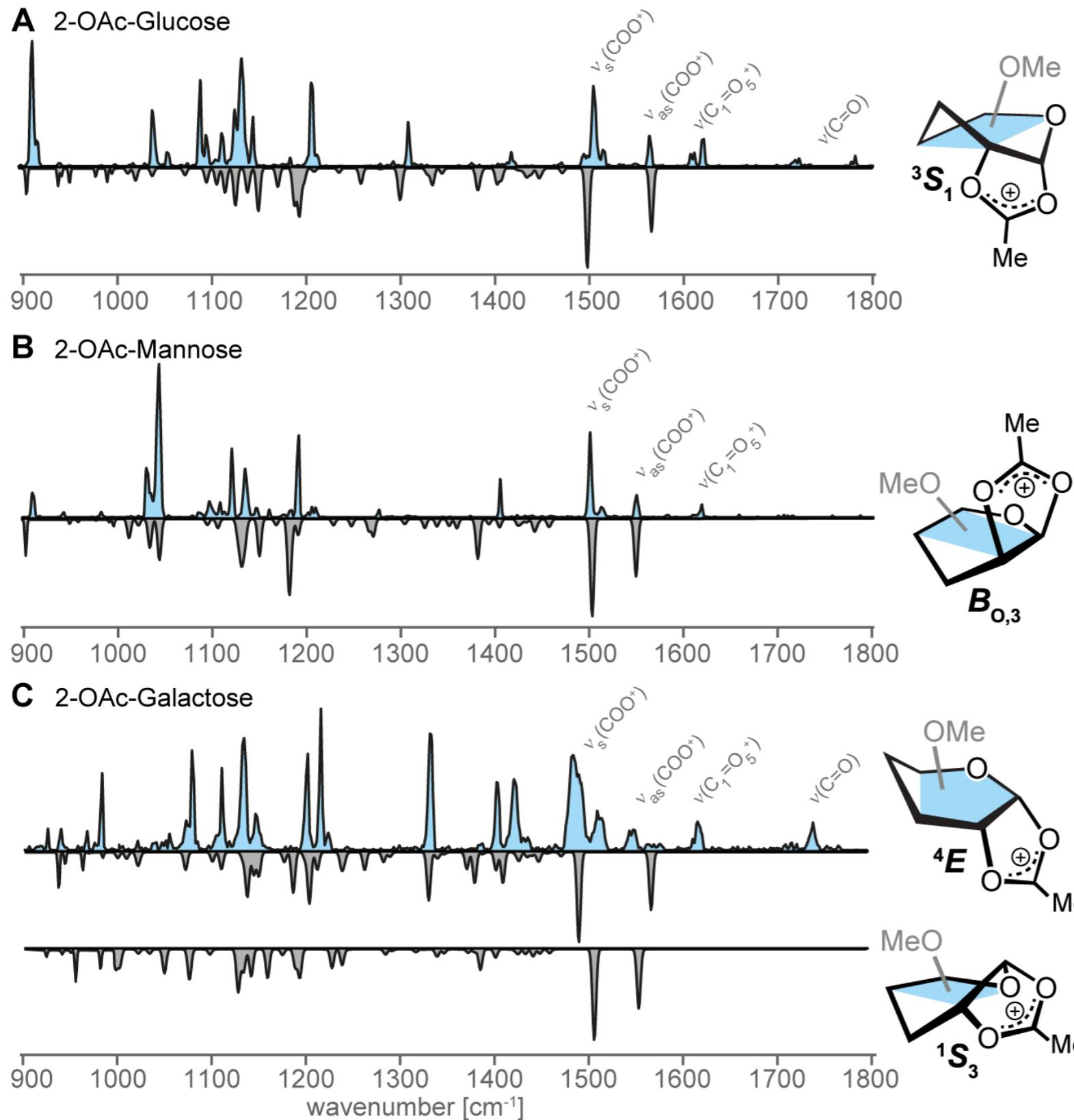
- Short-lived intermediate
- Exact structure unknown
- Too reactive to probe in condensed phase
- Can be isolated as a fragment in the gas phase



Probing the Structure of Glycosyl Cations

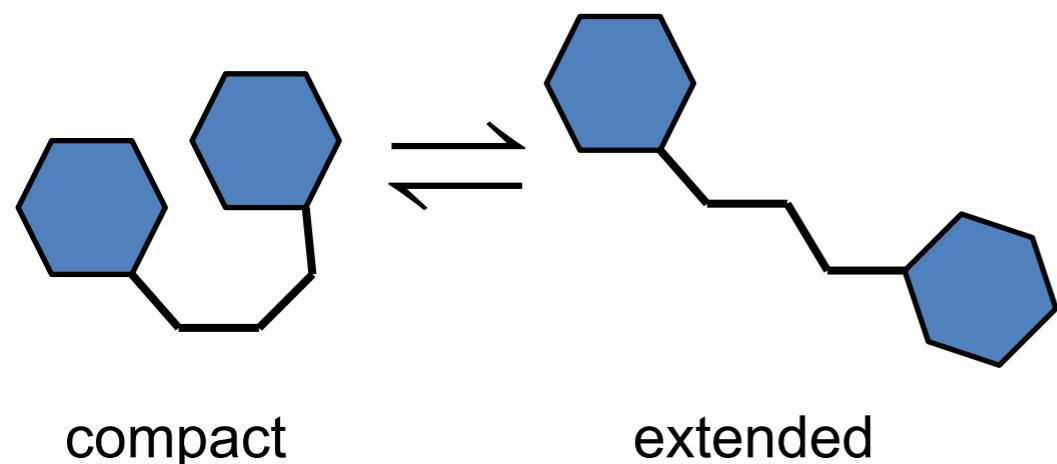


Probing the Structure of Glycosyl Cations

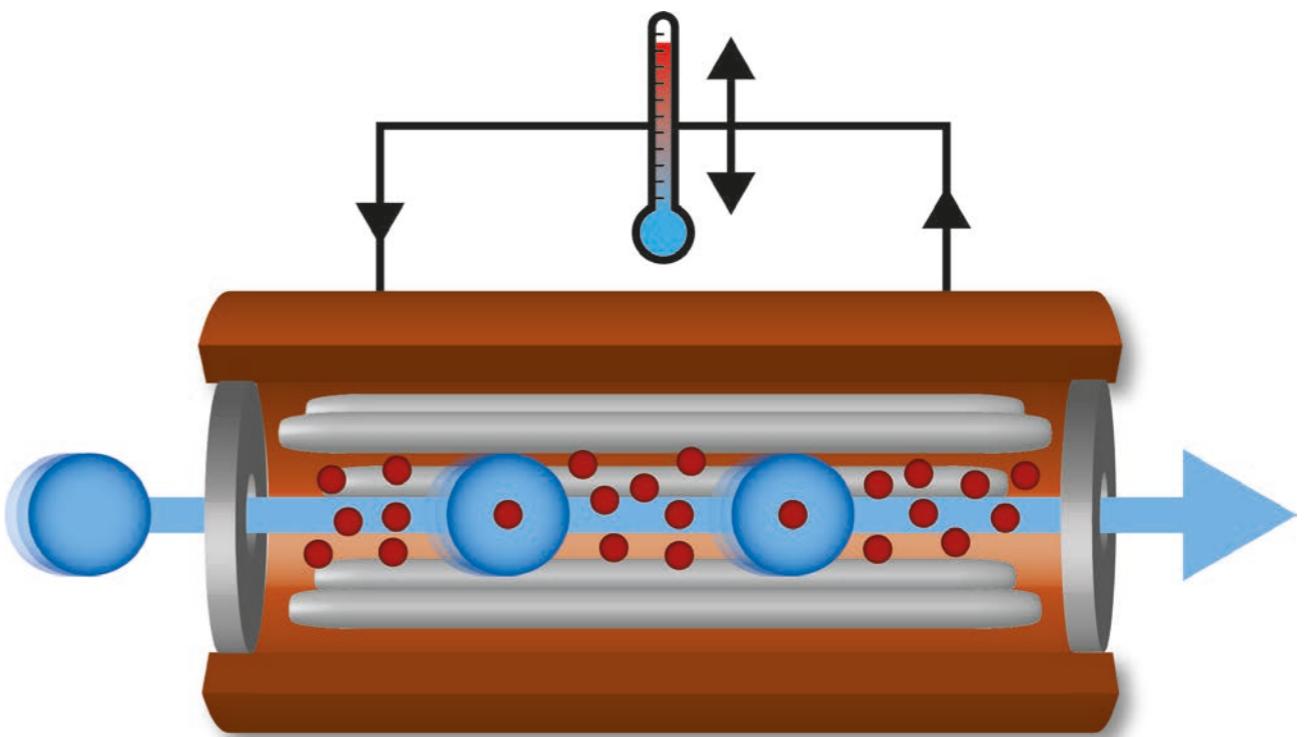


- Highly resolved IR spectra
- Theory provides low energy structure candidates
- Good agreement between theory and experiment
- Ions adopt dioxolenium structures
- Various ring pucksers found

Shock Freezing in Helium Nanodroplets



$$\ln K = \ln \frac{[\text{extended}]}{[\text{compact}]} = -\frac{\Delta H}{RT} + \frac{\Delta S}{R}$$

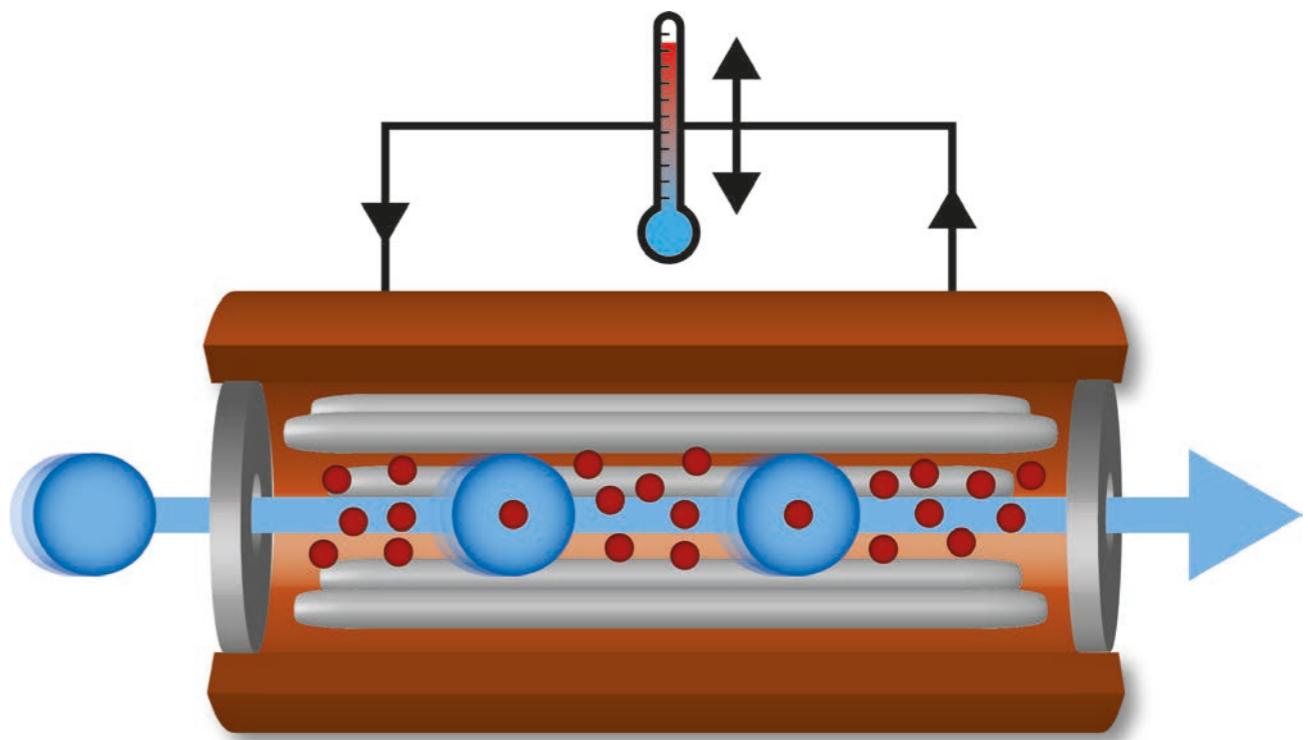
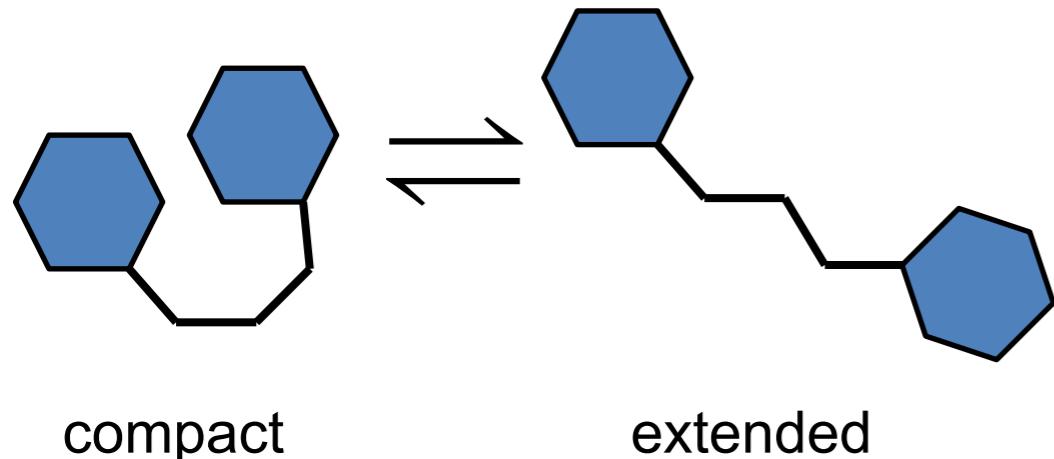


variable-temperature ion
trap (80-400 K)

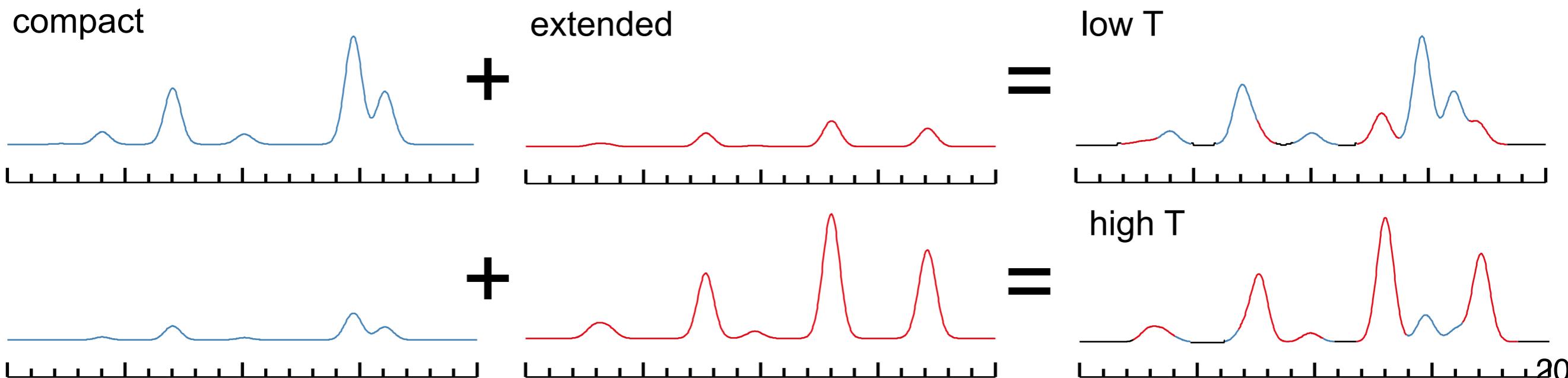
C. M. Leavitt *et al.*, *J. Phys. Chem. A* **2014**, *118*, 9692-9700.

D. S. Skvortsov, A. F. Vilesov, *J. Chem. Phys.* **2009**, *130*, 151101.

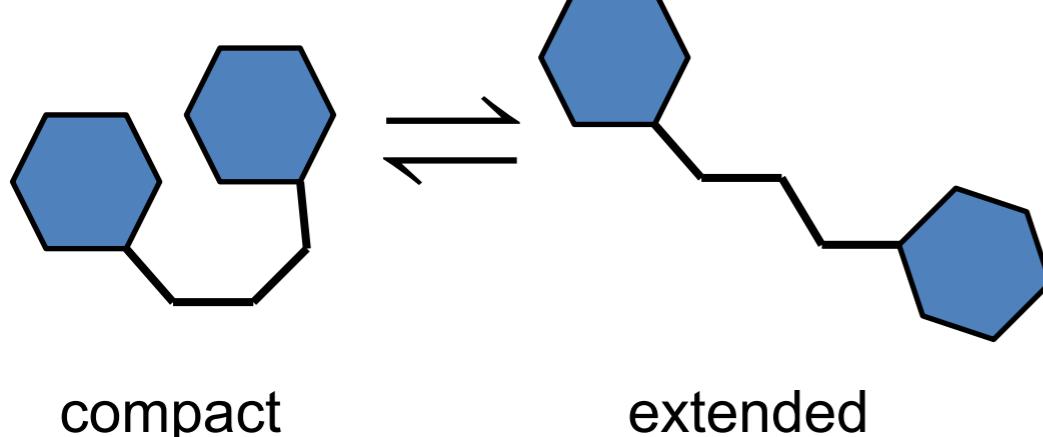
Thermochemical Measurements by HENDI IR



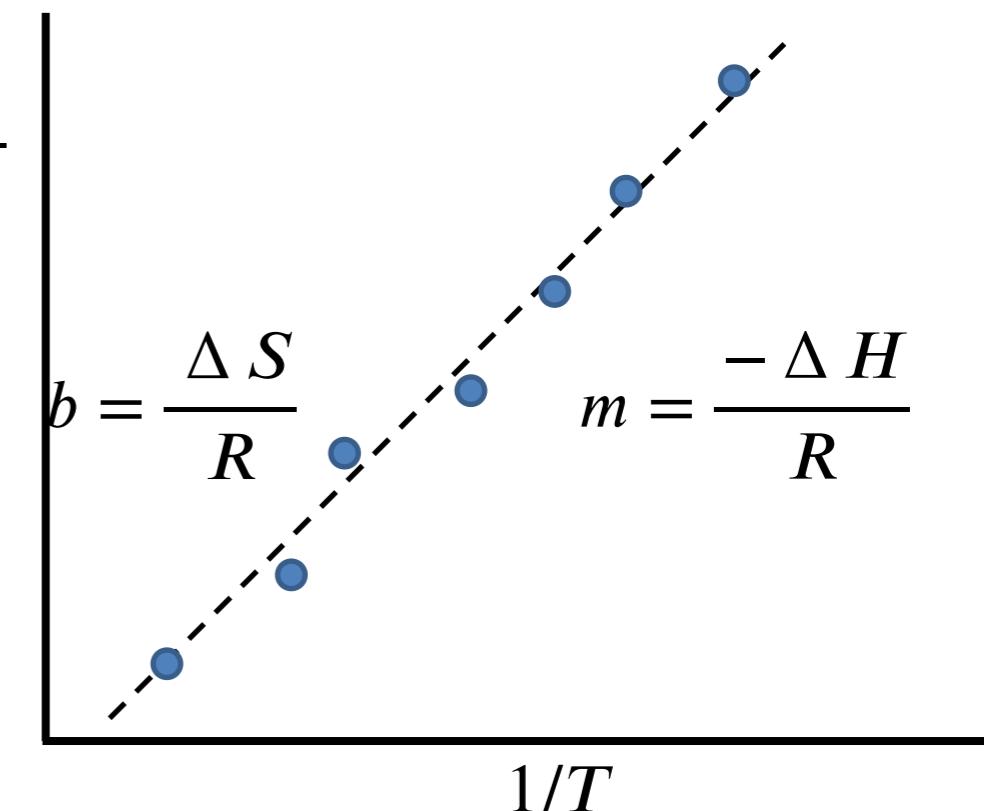
$$\ln K = \ln \frac{[\text{extended}]}{[\text{compact}]} = \frac{-\Delta H}{RT} + \frac{\Delta S}{R}$$



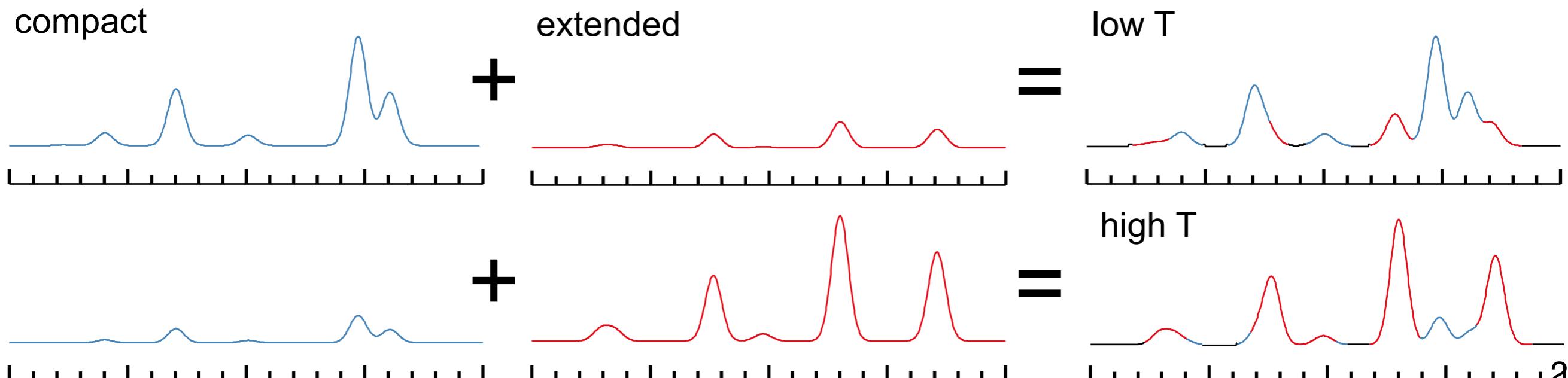
Thermochemical Measurements by HENDI IR



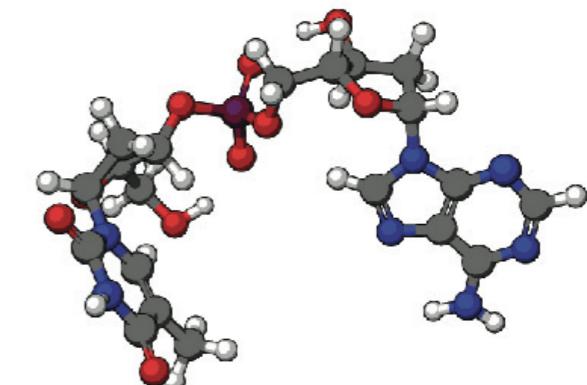
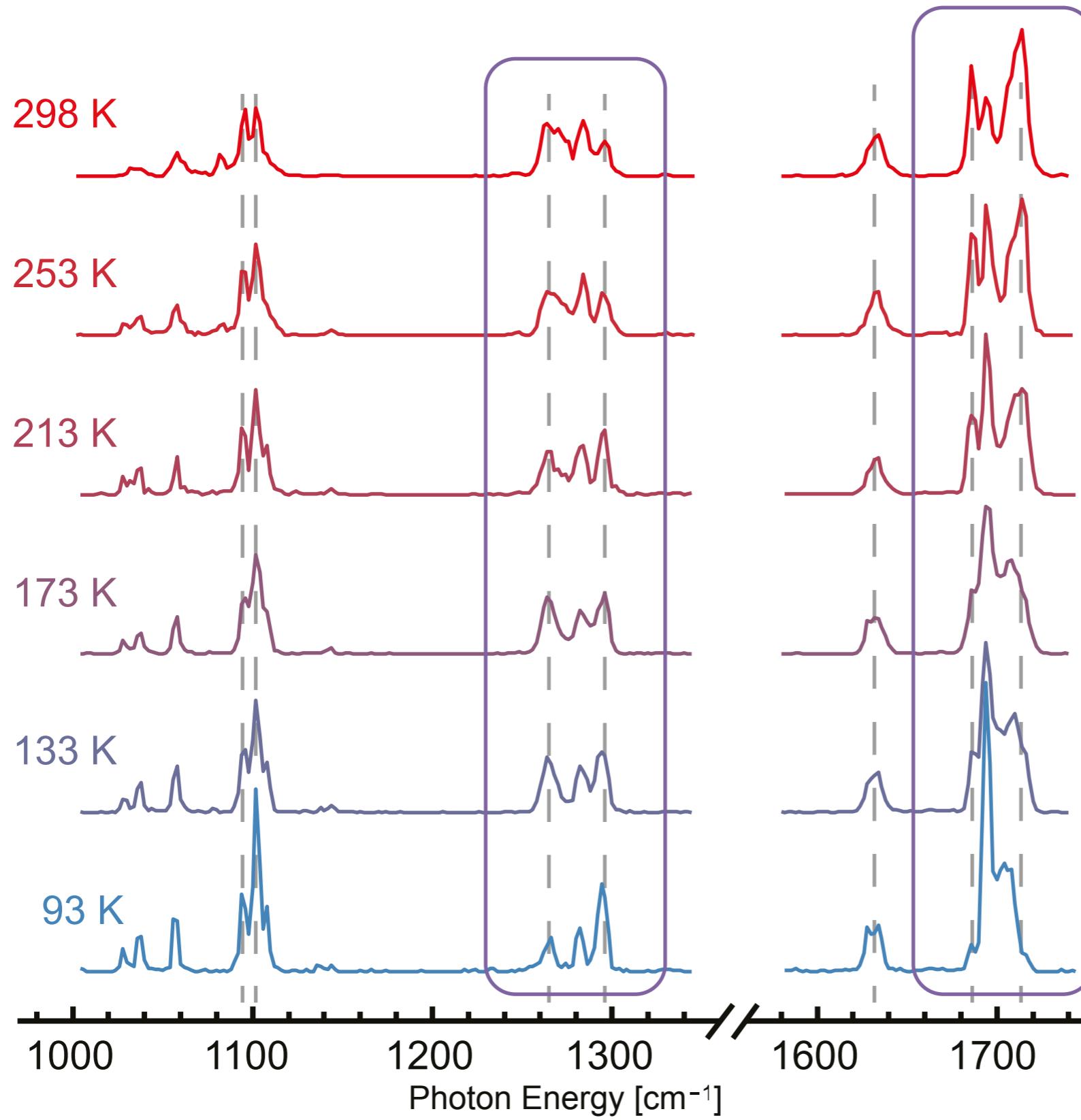
$$\ln \frac{[\text{extended}]}{[\text{compact}]}$$



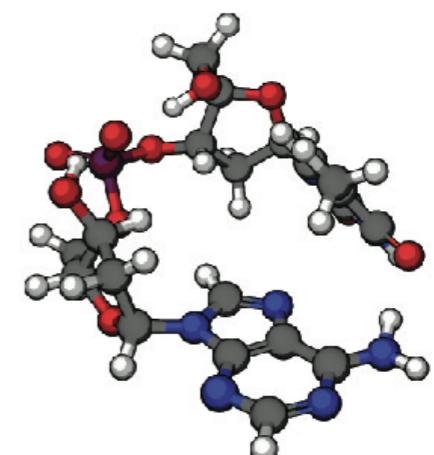
$$\ln K = \ln \frac{[\text{extended}]}{[\text{compact}]} = \frac{-\Delta H}{RT} + \frac{\Delta S}{R}$$



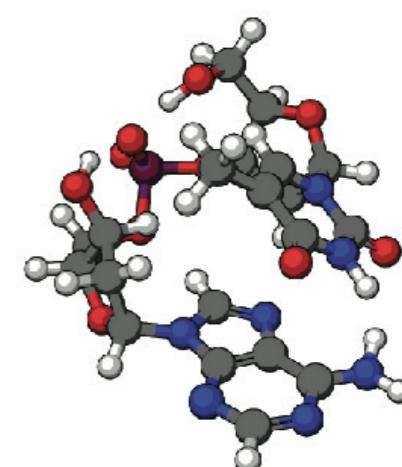
Proof of Principle – Dinucleotide dTp_dA



$$\Delta G_{298} = +3.3 \text{ kJ mol}^{-1} \quad (3)$$

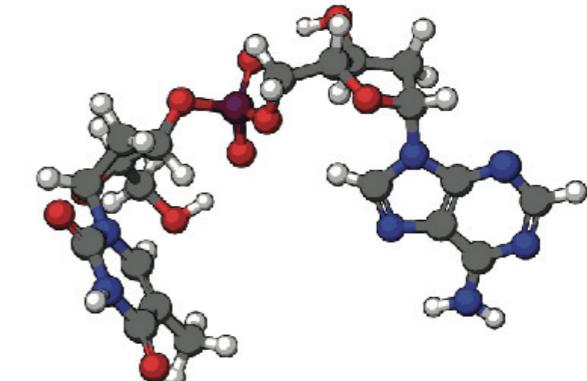
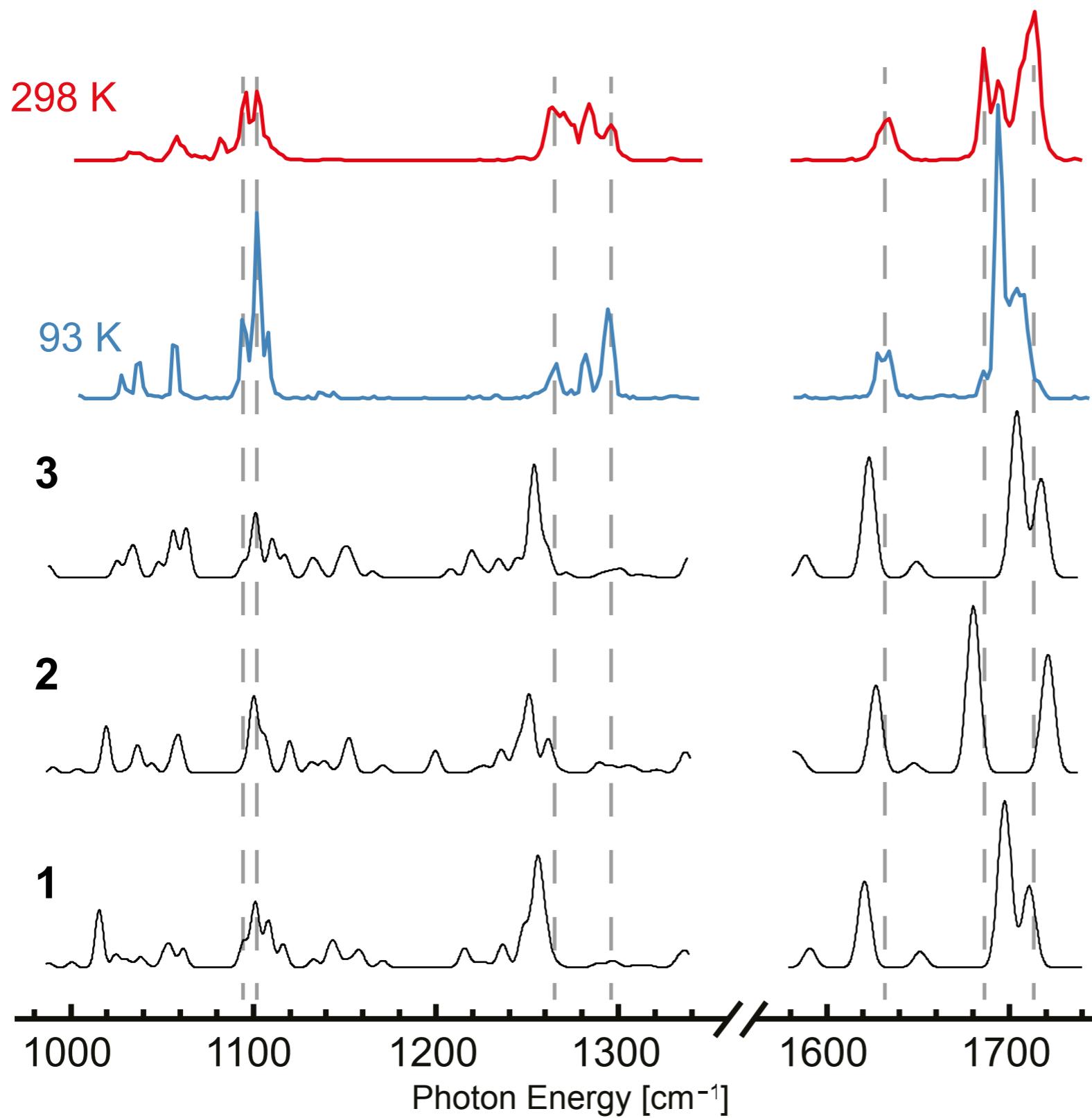


$$\Delta G_{298} = +1.5 \text{ kJ mol}^{-1} \quad (2)$$

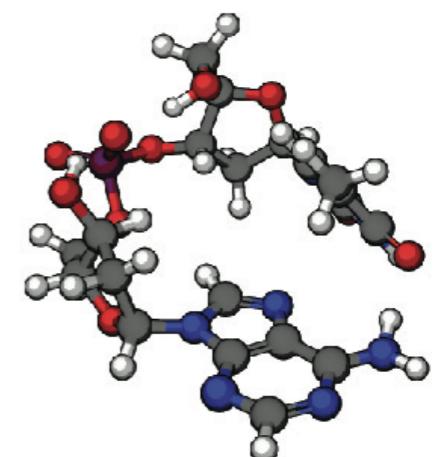


$$0.0 \text{ kJ mol}^{-1} \quad (1)$$

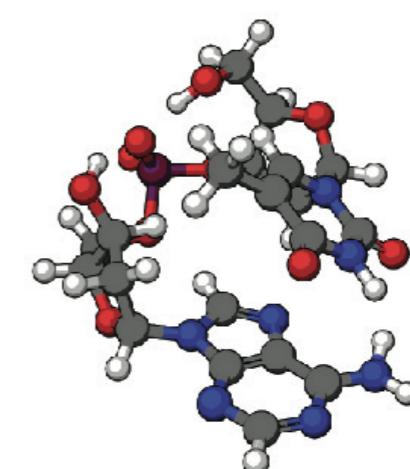
Proof of Principle – Dinucleotide dTp_dA



$$\Delta G_{298} = +3.3 \text{ kJ mol}^{-1} \quad (3)$$



$$\Delta G_{298} = +1.5 \text{ kJ mol}^{-1} \quad (2)$$



$$0.0 \text{ kJ mol}^{-1} \quad (1) \quad 23$$



Acknowledgements



MP:

Daniel Thomas
Eike Mucha
Mateusz Marianski
Gerard Meijer



FU Berlin / FHI:

Kevin Pagel



FEL Team:

Wieland Schöllkopf
Sandy Gewinner

Thank you!